

special destabilization of 7-norbornyl cations is much clearer in these data than it ever could be by considering solvolysis results, in which the effect of flattening at carbon is superimposed on the orbital symmetry effect.^{7,9}

Registry No. 1, 51689-29-3; 1⁺, 95484-94-9; 2, 88656-03-5; 2⁺, 95484-95-0; 3, 95484-96-1; 8-(bicyclo[3.2.1]octan-8-ylidene)bicyclo[3.2.1]octane, 95484-97-2; 9-(bicyclo[3.3.1]nonan-9-ylidene)bicyclo[3.3.1]nonane, 55993-21-0; 2-(2-adamantanylidene)adamantane, 30541-56-1; tetracyclo[6.2.1.1^{8,6}.0^{2,7}]tetradec-2,7-ene, 73321-28-5; 1,2,3,4,5,6,7,8,9,10,11,12,13,14-tetradecahydro-1,5:3,7:8,12:10,14-tetramethanononane, 30614-34-7.

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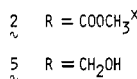
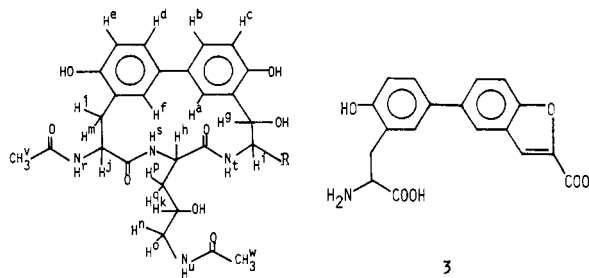
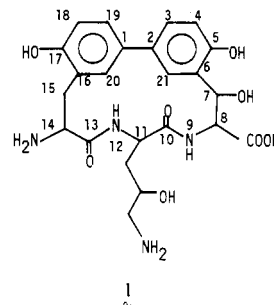
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Structure of WS-43708A, a Novel Cyclic Peptide Antibiotic

Summary: On the basis of chemical and spectroscopic evidence, the antibiotic WS-43708A (C₂₃H₂₈N₄O₈) has been shown to be a cyclic peptide (1) containing a biphenyl moiety included in a 15-membered ring.

Sir: WS-43708A (1), recently isolated from *Streptomyces griseorubiginosus* No. 43078, is a novel cyclic peptide with potent antibacterial activity.¹ Herein, we report the structure elucidation of this antibiotic on the basis of chemical and spectroscopic evidence.

WS-43708A was isolated as colorless needles from dilute HCl (pH 4): C₂₃H₂₈N₄O₈·2HCl (FABMS and elemental analysis);² mp 205–209 °C dec; [α]_D²⁰ -22.5° (c 0.1, 1 N HCl); IR (KBr) 3200, 3000–2300, 1690, 1640 cm⁻¹; UV (H₂O) 264 (ε 18600), 287 nm (sh); UV (0.1 N NaOH) 288 (ε 24000), 303 nm (sh); positive ninhydrin test. Acetylation of 1 with Ac₂O in MeOH (0 °C), followed by methylation with CH₂N₂ in MeOH (0 °C), gave the diacetyl monomethyl ester 2 (FABMS, *m/z* 587 (M⁺ + 1)). Hence, one carboxyl and two amino groups are present in 1. The ¹³C NMR spectrum (D₂O–DCl) of 1 showed in the sp³-carbon region eight signals including five methine signals attributable to two secondary alcohol carbons (64.4 (d) and 65.2 (d) ppm) and three α-amino acid carbons (50.9 (d), 55.0 (d), and 57.4 (d) ppm), the remainder being three methylene signals (30.4 (t), 37.9 (t), and 44.9 (t) ppm). In the sp²-carbon region 15 signals were assignable to three carbonyl groups (168.6 (s), 173.2 (s), and 174.0 (s) ppm) and two phenyl rings substituted totally with six substituents (116.4 (d), 116.9 (d), 120.3 (s), 126.2 (d), 127.2 (d), 127.6 (d), 127.9 (s), 130.6 (d), 132.9 (s), 133.0 (s), 152.8 (s), and



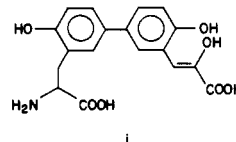
154.2 (s) ppm). Extensive spin decoupling (Table I) of the 400-MHz ¹H NMR spectra of 1 and 2 revealed ¹H–¹H relationships as shown in Figure 1, leading to partial structures A, B, and C, which are quite consistent with the ¹³C NMR data described above.

The partial unit A was confirmed by the fact that hydrolysis of 1 with 6 N HCl (110 °C, 24 h) gave, after chromatography on Toyopearl HW40S, *erythro*-γ-hydroxy-L-ornithine (HCl salt; mp 176–178 °C dec; [α]_D²³ +10.9° (c 1.0, H₂O)) which was identified by comparison with an authentic sample.³ The partial units B and C can be combined and extended to partial structure B + C (Figure 2) based on the following grounds. The acid hydrolysis described above also gave compound 3 (FDMS, *m/z* 341 (M⁺)),⁴ which was converted, by acetylation with Ac₂O in MeOH followed by treatment with CH₂N₂ in MeOH, to the monoacetyl trimethyl derivative 4 (high-resolution EIMS, *m/z* 425.1457, calcd for C₂₃H₂₃NO₇ 425.1472). The ¹H NMR analysis of 4 with the aid of decoupling (Table I) and NOE experiments (Figure 2) revealed the structure of 3. The genesis of 3 is rationalized by the following, plausible reaction mechanism from the partial units B and C: (1) dehydration of the β-hydroxy amino acid residue in C to the dehydro amino acid; (2) hydrolysis to the keto acid;⁵ (3) dehydrative condensation

(3) (a) Mizusaki, K.; Yamamoto, H.; Makizumi, H. *Bull. Chem. Soc. Jpn.* 1980, 53, 2605. (b) Mizusaki, K.; Makizumi, S. *Bull. Chem. Soc. Jpn.* 1981, 54, 470.

(4) ¹H NMR (CD₃OD–D₂O) δ 3.19 (dd, *J* = 7.5, 14 Hz, 1 H), 3.47 (dd, *J* = 5, 14 Hz, 1 H), 4.29 (dd, *J* = 5, 7.5 Hz, 1 H), 7.01 (d, *J* = 8.7 Hz, 1H), 7.49 (m, 2 H), 7.64 (s, 1 H), 7.66 (d, *J* = 8.7 Hz, 1 H), 7.70 (dd, *J* = 2, 8.7 Hz, 1 H), 7.90 (d, *J* = 2 Hz, 1 H).

(5) A subsequent inspection of the acid hydrolysate revealed the presence of keto acid i as a minor product, firmly supporting the mechanism leading to 3: ¹H NMR (CD₃OD–D₂O) δ 3.07 (dd, *J* = 8.8, 14 Hz, 1 H), 3.44 (dd, *J* = 4, 14 Hz, 1 H), 6.93 (d, *J* = 8.7 Hz, 1 H), 7.12 (s, 1 H), 7.32 (d, *J* = 8.7 Hz, 1 H), 7.42 (dd, *J* = 2.5, 8.7 Hz, 1 H), 7.49 (d, *J* = 2.5 Hz, 1 H), 7.59 (dd, *J* = 2.5, 8.7 Hz, 1 H), 7.67 (d, *J* = 2.5 Hz, 1 H).



(1) Umehara, K.; Ezaki, M.; Iwami, M.; Yamashita, M.; Hashimoto, S.; Komora, T.; Uchida, I.; Hashimoto, M.; Mine, Y.; Kohsaka, M.; Aoki, H.; Imanaka, H. "Abstracts of Papers", 24th Interscience Conference on Antimicrobial Agents and Chemotherapy, Washington, D.C., 1984, Abstr. No. 1141.

(2) FABMS, *m/z* 489 (M⁺ + 1) (free base); elemental analysis. Anal. Calcd for C₂₃H₂₈N₄O₈·2HCl·H₂O: C, 47.65; H, 5.62; N, 9.67; Cl, 12.23. Found: C, 47.80; H, 5.85; N, 9.75; Cl, 12.11.

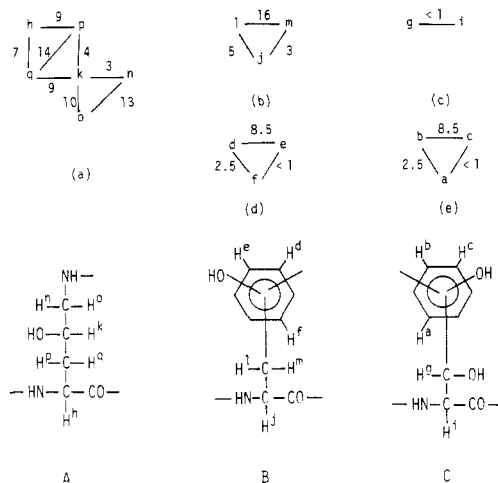


Figure 1. Partial structures A-C and the ^1H - ^1H relationships a-e. The ^1H - ^1H relationships a, b, and c were obtained by decoupling experiments on 1, while d and e were derived by those on 2. The very small vicinal coupling constant of the H^e - H^i is presumably due to a dihedral angle close to 90° caused by a restricted conformation of WS-43708A.

Table I. ^1H NMR (400 MHz) Chemical Shifts, Multiplicities, and Coupling Constants (J , Hz) for WS-43708A (1), 2, and 4

	1 ^a	2 ^b	4 ^c
H ^a	7.40, m (3 H)	7.39, d (2.5)	7.78, br d (2)
H ^b and H ^c	6.93, m (2 H)	6.83, d (8.5)	7.61, dd (2, 8.5)
H ^d		6.83, d (8.5)	7.62, br d (8.5)
H ^e		7.11, dd (2.5, 8.5)	7.48, dd (2.5, 8.5)
H ^f	6.87, br s	6.78, d (8.5)	6.96, d (8.5)
H ^g	5.84, br s	6.88, d (2.5)	7.33, d (2.5)
H ^h	5.02, dd (7, 9)	5.70, br s	7.57, s
H ⁱ	4.91, br s	5.09, dt (8.8, 7.5)	
H ^j	4.47, dd (3, 5)	4.63, br d (9.5)	
H ^k	4.09, dddd (3, 4, 9, 10)	4.56, dt (7.5, 3.3)	4.80, dt (7.5, 7.5)
H ^l	3.55, dd (5, 16)	3.64, m	
H ^m	3.03, dd (3, 16)	3.16 ^d	3.22, d (7.5)
H ⁿ	3.17, dd (3, 13)	3.16 ^d	
H ^o	2.97, dd (10, 13)	3.07, m	
H ^p	2.11, ddd (4, 9, 14)	1.79, m	
H ^q	1.95, ddd (7, 9, 17)	1.50, m	
H ^r		7.62, d (7.5)	6.23, d (7.5)
H ^s		8.50, d (8.8)	
H ^t		8.53, d (9.5)	
H ^u		7.88, t (5.5)	
H ^v		1.84, s (6 H)	1.95, s (3 H)
H ^w			
H ^x		3.71, s (3 H)	4.10, s (3 H)
H ^y			3.73, s (3 H)
H ^z			3.91, s (3 H)

^aD₂O-DCl. ^bMe₂SO-*d*₆. ^cCDCl₃. ^dOverlapping signals of H^l and Hⁿ prevented the examination of their multiplicities.

with the phenolic hydroxy group to the benzofuran structure 3. These chemical evidence thus leads to the partial structure B + C.

The problem remaining is to link A and B + C for the full structure of WS-43708A. Since a molecular model study indicated that an intramolecular cyclization of B + C itself is practically impossible, insertion of fragment A was proposed. This was corroborated by the ^1H NMR spectrum of 1 in D₂O-NaOD, in which, in contrast to up-field shifts of 0.36 and 0.67 ppm, respectively, for Hⁱ and H^j (δ 4.90 and 4.47 in D₂O-DCl), no shift was observed on

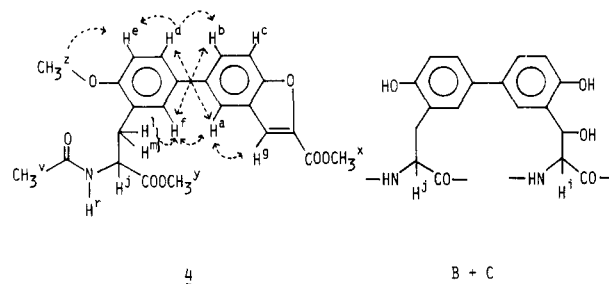


Figure 2. Structure of 4 and the partial structure B + C. The observed NOEs in 4 are shown by dotted line arrows.

H^h (δ 5.02 in both D₂O-DCl and D₂O-NaOD), indicating that the α -amino acid group of A is incorporated in a cyclic peptide structure.^{6,7} Reduction of 2 with NaBH₄ in MeOH gave alcohol 5 (FABMS, m/z 559 ($M^+ + 1$)), in the ^1H NMR spectrum (CD₃OD) of which a singlet-like signal (δ 4.89, CD₃OD) corresponding to Hⁱ in 2 was changed to a triplet (δ 4.20, $J = 7.5$ Hz) coupled to the newly formed methylene group (δ 3.65 (dd, $J = 7.5, 11$ Hz) and 3.78 (dd, $J = 7.5, 11$ Hz)). This indicates that the carboxylic acid function in 1 is bonded to C-8 bearing Hⁱ. Consequently, peptide bonds between N-9/C-10 and N-12/C-13 are postulated to give the full structure of 1, without stereochemistry, for WS-43708A.⁸

The biphenyl group incorporated in the 15-membered cyclic peptide portion of WS-43708A is unique and such compounds are rarely found in nature.⁹ The exceptional activity of WS-43708A against gram-positive bacteria will be reported separately.

Registry No. 1, 95485-50-0; 2, 95485-51-1; 3, 95485-52-2; 4, 95485-53-3; 5, 95485-54-4; erythro- γ -hydroxy-L-ornithine hydrochloride, 95485-55-5.

(6) For the pH-dependent chemical shifts of the α -methine protons of peptides, see: Sheinblatt, M. *J. Am. Chem. Soc.* 1966, 88, 2845.

(7) The ^1H NMR spectrum (Me₂SO-*d*₆) of 2 showed a triplet ($J = 5.5$ Hz) corresponding to the amido H originated from the δ -amino group of A at δ 7.88 coupled to Hⁿ and H^o (δ 3.07 (m) and 3.16 (m)), thus supporting that the δ -amino group of A in 1 is unsubstituted.

(8) Stereochemical deductions aside from that of the ornithine portion are the subject of future publications.

(9) As far as we are aware, the only antibiotics that contain a biphenyl group as part of a cyclic peptide system are those that belong to the vancomycin group of antibiotics, see: Barna, J. C. J.; Williams, D. H. *Annu. Rev. Microbiol.* 1984, 38, 339.

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Total Synthesis of (-)-Gilmicolin

Summary: The total synthesis of (-)-gilmicolin is disclosed; assignment of absolute stereochemistry is thereby secured.